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CHARACTERIZATION OF $\text{Co}_x\text{Zn}_{7-x}\text{Sb}_2\text{O}_{12}$ SPINEL OBTAINED BY PECHINI METHOD

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Abstract

Oxides with spinel structure have been studied for many decades. The reason is a wide variety of application as pigments, refractories and also as a model structure to studies that evaluate the relative stability of ions in octahedral and tetrahedral sites. Zinc-antimonium spinels ($\text{Zn}_7\text{Sb}_2\text{O}_{12}$) were synthesized by Pechini method, substituting the cation Zn^{2+} by Co^{2+} , according to the stoichiometry $\text{Co}_x\text{Zn}_{7-x}\text{Sb}_2\text{O}_{12}$ ($x = 0 - 7$). Thermal characterization of the powders, after pyrolysis, was done using TG/DTA analysis, being observed different mass losses, according to the cobalt amount in the resin. Powders were calcined at 600 to 1000 °C and characterized by XRD, BET and IV spectroscopy. Maximum cristalinity was obtained at 1000 °C. Materials with $x = 5 - 7$ presented a secondary phase $(\text{Co}, \text{Zn})\text{Sb}_2\text{O}_6$. Using infrared analysis, the possible sites where Zn^{2+} , Co^{2+} and Sb^{2+} are located in the net were inferred.

Keywords: spinel, $\text{Zn}_7\text{Sb}_2\text{O}_{12}$, Pechini, cobalt

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