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LOCAL STRUCTURE OF A NEW FAMILY OF HIGH BaO CONTENT GLASSES BASED ON BaO-SiO₂-Al₂O₃ SYSTEM

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Palabras Clave: Glass Phase diagrams MAS-NMR FT-IR

Resumen:

Two glasses labeled BAS-6, and BAS-7 based on the BaO-SiO₂-Al₂O₃ equilibrium diagram located within the compatibility triangle (BaSiO₃-Ba₂SiO₄-BaAl₂Si₂O₈) were fabricated. The main criteria used to select the glass composition was the precipitation after annealing (850°C, 2h) of Barium Silicate compounds and Hexacelsian due to its relatively high Coefficient of Thermal Expansion (CTE), 10–12x10⁻⁶/°C and up to 8x10⁻⁶/°C respectively, and its excellent dielectric properties. These properties lead to the potential application of these obtained glass-ceramics in a large panoply of added value fields, such as sealings, coatings and optical applications. In these compositions, the amount of BaO (R⁺⁺ glass modifier) ranges around 70 wt% and Al₂O₃ content around 5% wt. In order to reach adequate viscosity to cast the melt glass, part (5-10wt%) of the glass-forming SiO₂ was replaced by B₂O₃. The structure of the glasses was studied by Magic Angle Spinning Nuclear Magnetic Resonance (MAS-NMR) and Fourier Transform Infrared Spectroscopy (FTIR). It was indicated by Si-29 MAS-NMR spectra that silicon was present in different coordination with the Oxygen. Al-29 MAS-NMR spectra showed that four-fold coordinated aluminum Al (IV) was the dominant species with a second peak assigned to octahedral aluminum Al (VI). The FT IR and 11-B spectra suggested that the boron oxide has a large influence on the glass structure because the conversion of boron from triangular to tetrahedral coordination increases the connectivity of the structure. Furthermore, a distribution of silicate network including Si-O-Si stretching Q(4) and Q(3) and Si-O-NBO, Q(3) per SiO₄ was observed by the FTIR spectra.