CAT-07 THE ELECTRO-OXIDATION OF ISOTOPICALLY LABELED GLYCEROL AS A PROOF OF THE CLEAVAGE OF C-C CHAIN ON PT SURFACES

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This work presents the first in situ FTIR study with isotopically labeled glycerol on polycrystalline platinum and reveals new information concerning the glycerol C-C bond cleavage and the behavior of $13\mathrm{CO}_2$ and $12\mathrm{CO}_2$ as electro-oxidation products. The electro-oxidation of isotopically labeled glycerol generates both $13\mathrm{CO}_2$ and $12\mathrm{CO}_2$ and indicates that glycerol is able to dissociate on Pt. The oxidation of $-13\mathrm{CH}_2\mathrm{OH}$ is easier than that of the central group. Results are interpreted in terms of a more favorable position of $-13\mathrm{CH}_2\mathrm{OH}$ groups to react with Pt-OH $_{\mathrm{ad}}$ species.

CAT-08 PREPARAÇÃO DE ELETROCATALISADORES PtAuSn/C VIA REDUÇÃO POR FEIXE DE ELÉTRONS PARA A OXIDAÇÃO ELETROQUÍMICA DO ETANOL

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Palavras-chave: PEMFC, etanol, eletrocatalisadores, Pt. Au. Sn. oxidação

PtAuSn/C electrocatalysts (20 wt.% metal loading) were prepared in water/2-propanol using electron beam irradiation. The diffractograms of the PtSn/C and PtAuSn/C electrocatalysts showed peaks associated to Pt face-centered cubic structure. The materials were tested in the electro-oxidation of ethanol in an acidic PEMFC. The activity of the electrocatalysts for alcohol electro-oxidation in acid medium showed that PtAuSn/C electrocatalysts had a higher performance than PtSn/C E-TEK commercial.