

### OB13- PERSISTENT LUMINESCENCE OF BARIUM ALUMINATE DOPED WITH $\text{Eu}^{2+}$ PREPARED BY THE COMBUSTION METHOD

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Persistent luminescent materials containing  $\text{Eu}^{2+}$  have been studied extensively in the last years. The  $\text{MAl}_2\text{O}_4:\text{Eu}^{2+},\text{Dy}^{3+}$  ( $\text{M} = \text{Ca}, \text{Sr}, \text{Ba}$ ) materials are usually prepared by ceramic methods using annealing at high temperatures (about 1500 °C) for a long time in a reducing atmosphere. In order to obtain  $\text{BaAl}_2\text{O}_4:\text{Eu}^{2+},\text{Dy}^{3+}$  at low temperature, without a reducing atmosphere and with nanoscale particles, the combustion method was used with urea as the fuel. The material was efficiently prepared at 400 °C without a reducing atmosphere and was characterized by X-ray powder diffraction, Scanning Electron Microscopy (SEM), infrared absorption spectroscopy as well as with luminescence spectroscopy with life time and luminance measurements.

The analyses demonstrated the formation of the  $\text{BaAl}_2\text{O}_4$  material evidenced by the X-ray pattern (Fig. 1) with efficient reduction of  $\text{Eu}^{3+}$  to  $\text{Eu}^{2+}$  as shown by the luminescence spectra (Fig. 2). The SEM image is shown in Fig. 3.

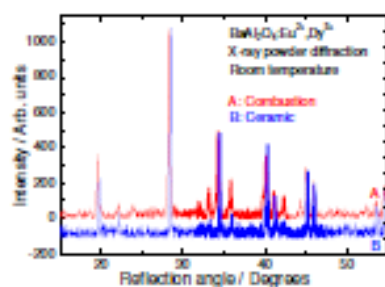


Fig. 1: X-ray patterns of the material prepared by a) combustion and b) ceramic methods.

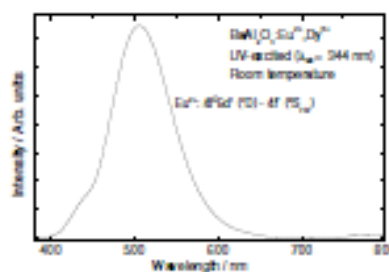


Fig. 3: SEM image of  $\text{BaAl}_2\text{O}_4:\text{Eu}^{2+},\text{Dy}^{3+}$ .

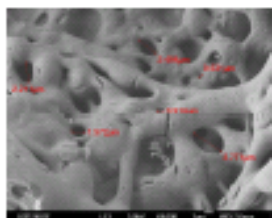


Fig. 2: Luminescence spectrum of  $\text{BaAl}_2\text{O}_4:\text{Eu}^{2+},\text{Dy}^{3+}$  with UV-excitation.