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Programme and
Book of Abstracts

MC-02

NANOMATERIALS SYNTHESIS BY BENZENETRICARBOXYLATE METHOD

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With improvements made lately in the nanoscience and nanotechnology, the photonic, structural and morphological properties of the RE₂O₃ doped with Eu³⁺ nanomaterials (RE₃₊: Y, Gd and Lu) have been widely investigated [1]. In the thermolysis route, it is of great advantage to use the [RE(TLA)] complexes as precursors (TLA: 1,2,4-benzenetricarboxylate) to produce the cubic RE₂O₃ materials (500–100°C). The thermogravimetric curves of the [RE(TLA):Eu (x mol%)] complexes (RE₃₊: Y, Gd and Lu; x = 0.1, 0.5, 1.0 and 5.0) shows that the organic moiety decomposes in a single-step from 430 to 580 °C, leading to the formation of RE₂O₃ at 500 °C [2]. The decomposition temperature decreases for all complexes with increasing Eu³⁺ concentration. The excitation e emission spectra of the systems present the characteristic emission of the Eu³⁺ ion. The excitation spectra show a broad oxide to europium charge transfer absorption band at 260 nm, i.e. O₂(2p)@Eu₃₊(4f₆) LMCT, as well as the narrow lines assigned to the 4f₆ intraconfigurational transitions of the Eu³⁺ ion. The increase of the annealing temperature decreases the value of the Ω₂ intensity parameter. We believe that the Eu³⁺ could migrate from the C₂ sites (non-centrosymmetric) to S₆ (centrosymmetric) at higher temperatures reducing the overall Ω₂ value. The optical results are consistent with the low symmetry of the C₂ site occupied by the Eu³⁺ ion in the cubic C-type RE₂O₃:Eu³⁺, which is essential for optical applications. In conclusion, the Y₂O₃:Eu³⁺ nanophosphors exhibit high values of emission quantum efficiency, compatible with the commercial phosphors currently available in the photonic

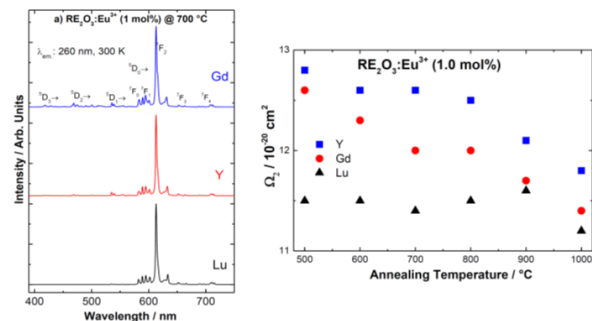


Figure 1. a) Emission spectra of RE₂O₃:Eu³⁺ (1.0 mol%) materials (RE₃₊: Y, Gd and Lu) and b) variation of the Ω₂ experimental intensity parameter with the annealing temperature.

References

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MATERIALS

MC-03

CHEMICAL AND HYDROSTATIC PRESSURE EFFECTS ON SPECTROSCOPIC PROPERTIES OF Nd³⁺-DOPED GALLIUM NANO-GARNETS

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Matter under extreme conditions of pressure is the subject of multidisciplinary studies that join such diverse fields as physics, chemistry, material science, geology, microbiology, or food technology. One of the most interesting goals of the high pressure technique is that provides a unique insight into the electronic structure and optical properties of materials doped with rare earth (RE³⁺) ions, since it can vary the structure, the coordination environment and, consequently, the electric, magnetic, vibrational, and optical properties.¹ On the other hand, the use of the trivalent Neodymium (Nd³⁺) ion-doped inorganic materials have attracted great attention in many areas, from science research to industry applications. Garnets have always been attractive gain media for laser applications due to their high mechanical strength and good thermal and optical properties. In this sense, much efforts have been spent in the last decade in the study of the luminescence properties of RE³⁺-doped Gd₃Ga₅O₁₂ (GGG), Y₃Ga₅O₁₂ (YGG) and Lu₃Ga₅O₁₂ (LuGG) nanocrystalline garnets, especially as an alternative to quantum dots in photonic devices.

We propose the comparative study of the effects of chemical and hydrostatic pressures on the structural, vibrational and optical properties of the Nd³⁺ ion in a series of RE₃Ga₅O₁₂ rare earth gallium crystalline and nano-crystalline garnets when the size of the RE³⁺ (=Gd³⁺, Y³⁺, Lu³⁺) changes. Pressure will modulate the structural, vibrational and elastic properties of the garnet. In addition, the crystal-field interaction felt by the RE³⁺ ions is expected to be different in these three host lattices due to the magnitude of the size mismatch with the ligands. The size effects also generate stresses, strains and rearrangements in the first coordination sphere of the Nd³⁺ optically active ion. Further modifications in the free ion interactions and/or in the crystal-field, and hence, in the optical properties are then induced when hydrostatic pressure is applied.

References

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