

# Density Functional Theory study of Cd impurities in Molybdenum Trioxide

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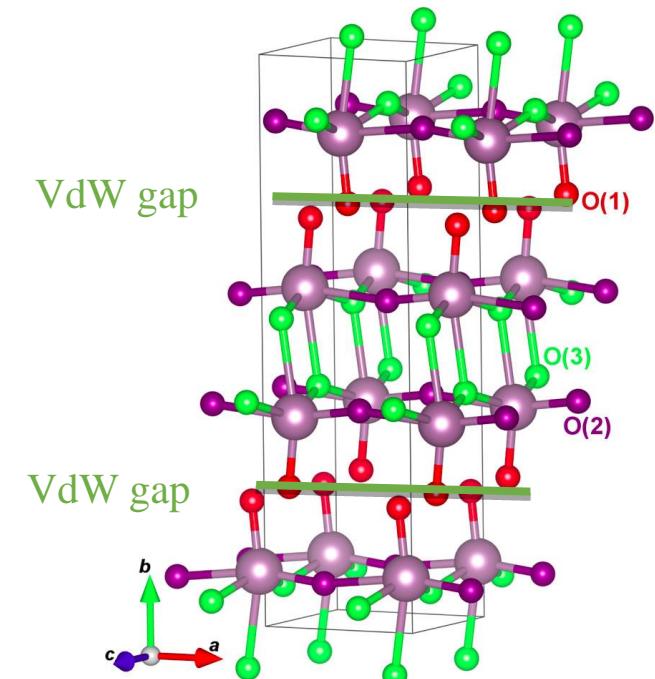
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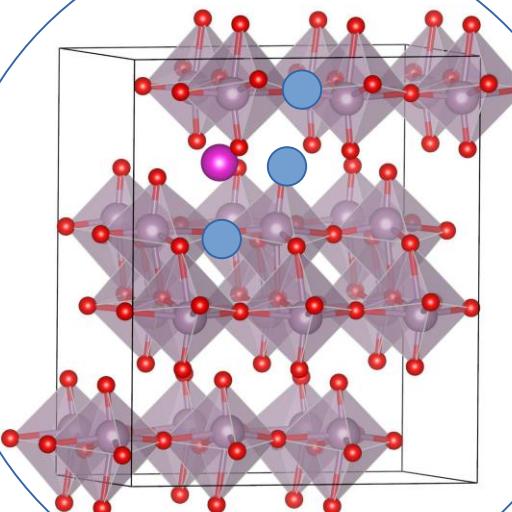
## Introduction

- The molybdenum trioxide ( $\text{MoO}_3$ ) is known for its photo-, thermo- and electro-chromism, high catalytic activity.
- the  $\alpha\text{-MoO}_3$  is the most stable crystal phase and it has a layered structure consisting of van der Waals bonded sheets of distorted edge-sharing  $\text{Mo}-\text{O}_6$  octahedra in which Mo atoms are bounded by three distinct types of oxygen atoms.

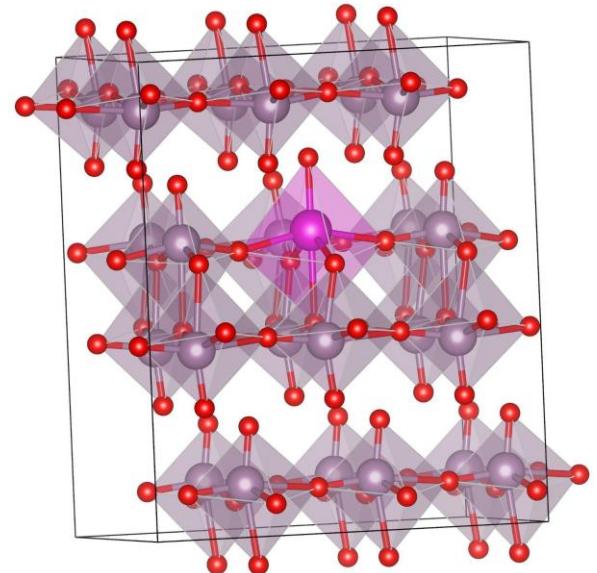


The hyperfine parameters of  $^{111m}\text{Cd}(^{111}\text{Cd})$  impurities in  $\alpha\text{-MoO}_3$  are investigated by first-principle calculations in the framework of density functional theory (DFT). The Generalized gradient approximation (GGA) plus Hubbard-U corrections for onsite Coulomb interactions are used in the DFT calculations. Also in the calculations, the effect of van der Waals forces between layers are considered.

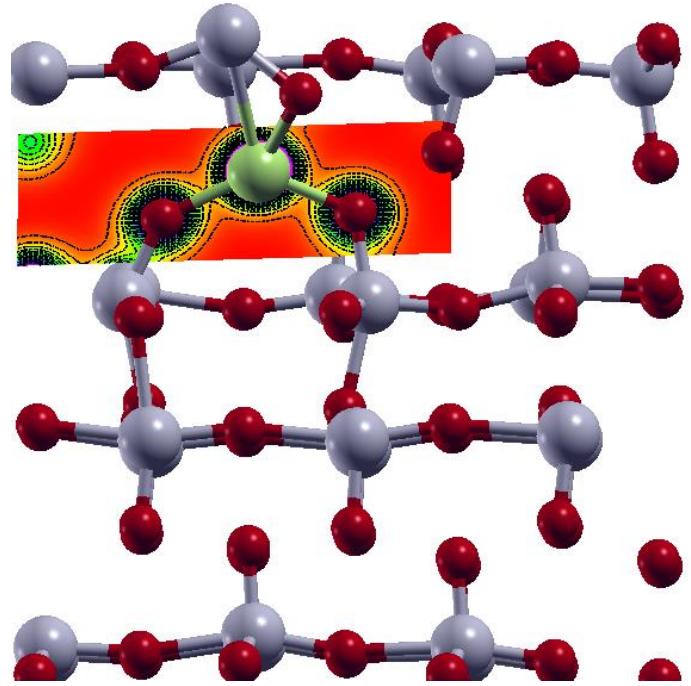
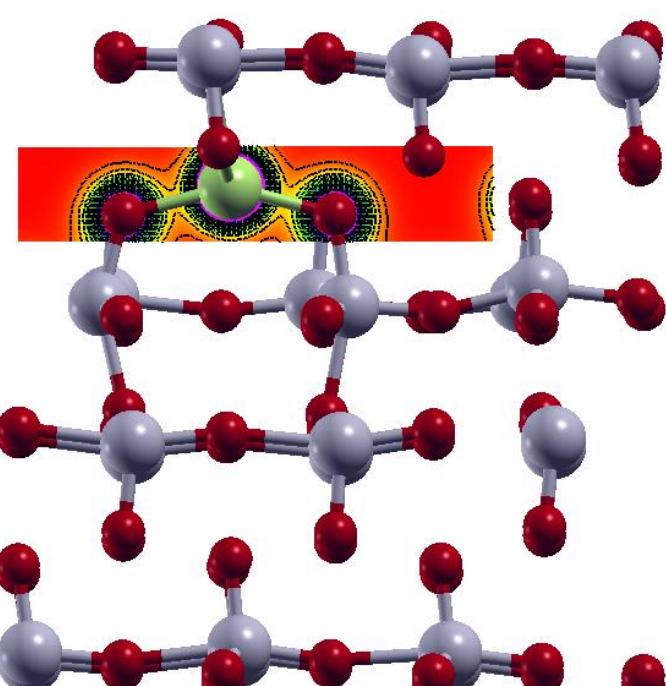
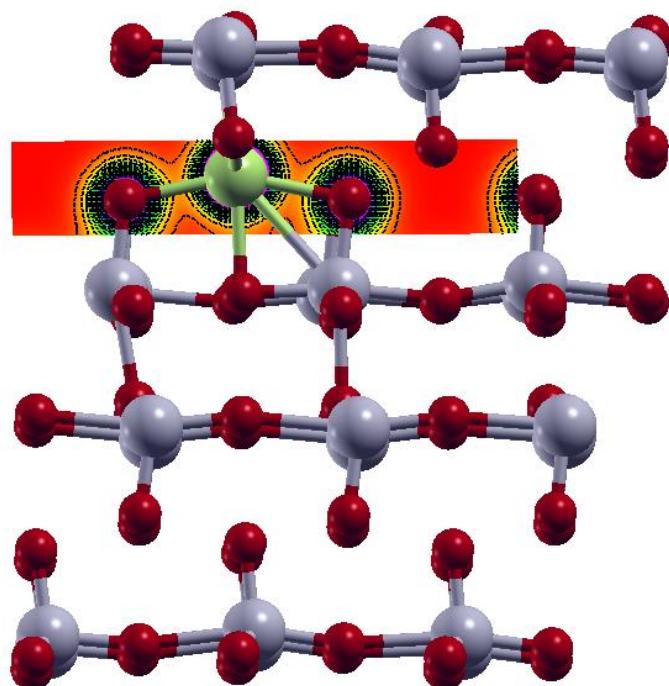
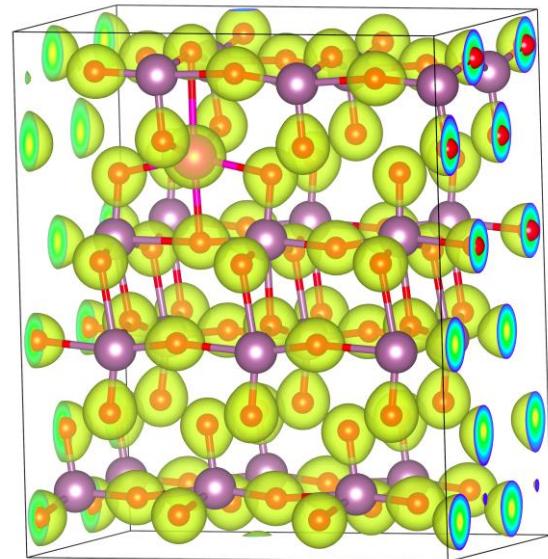
Approximation	a(Å)	%	b(Å)	%	c(Å)	%
GGA-PBE	3.9495	0.34%-	13.8086	0.33%-	3.684	0.34%-
LDA	4.002	0.99%	13.9921	0.99%	3.733	0.99%
PBE-Vdw	3.976	0.15%	13.901	0.33%	3.708	0.05%
.Exp	3.97		13.85		3.71	



PBE + SP + U = 6eV (Mo)			
State	Vzz (V/Å²)	$\eta$	E_F (eV)
Cd <sub>I</sub>	54.13	0.240	1.3203
Cd <sub>S</sub>	45.18	0.957	7.8493



Hyperfine parameters			Electron configuration /Charge state
Configuration	Vzz (V/Å <sup>2</sup> )	η	d
3x1x2 / GGA+U	54.13	0.240	[Kr]4(s <sup>2</sup> p <sup>6</sup> )4d <sup>8</sup> /Cd <sup>4+</sup>
3x1x2 / GGA+U + 1e	54.98	0.177	[Kr]4(s <sup>2</sup> p <sup>6</sup> )4d <sup>8</sup> /Cd <sup>3+</sup>



Approximation: GGA/PBE with VdW D3 functional				Experimental			
	PBE + U = 6eV (Mo)						
Configuration	V <sub>zz</sub> (V/Å <sup>2</sup> )	η	EFG	V <sub>zz</sub>   (V/Å <sup>2</sup> )	η	%	°C
Cd <sub>I</sub>	54.98	0.177	EFG2	68(1)	0.19(3)	24(3)	24
Cd <sub>I</sub> <sup>VO1</sup>	134.26	0.086	-	-	-	-	-
<sup>VO1</sup> Cd <sub>I</sub> <sup>VO1</sup>	185.17	0.215	-	-	-	-	-
Cd <sub>I</sub> <sup>VO2</sup>	190.24	0.373	EFG 3	195(2)	0.32(1)	13(2)	24
<sup>VO2</sup> Cd <sub>I</sub> <sup>VO2</sup>	48.87	0.255	-	-	-	-	-
<sub>VO2</sub> Cd <sub>I</sub> <sup>VO2</sup>	172.51	0.547	EFG 4	163(1)	0.484(4)	37(3)	150

## Conclusion

- After implantation, Cd ion is located at interstitial position in MoO<sub>3</sub> and we find that a vacancy is most likely to form on a 2-fold coordinated O(2) atom.
- the conversion of a former single polaron at Cd<sub>I</sub> into a bi-polaron state is happened by increasing the annealing temperature