

Lanthanum-ceria modified with calcium for oxidative coupling of methane

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Ethylene (C₂H₄) is considered a fundamental component of the chemical industry, it is a raw material widely used to produce plastics, solvents, fertilizers, etc. Currently, the main route to obtain C₂H₄ is the steam cracking of naphtha from crude oil, a well-established technology that is hardly challenged. The search for more sustainable alternatives must be ongoing, and among the alternative, we can mention the use of natural gas. The oxidative coupling of methane (OCM), the main component of natural gas, is a compelling direct pathway of methane (CH₄) conversion into higher added value hydrocarbons, such as C₂(C₂H₆ and C₂H₄). [1] La₂Ce₂O₇ catalyst performs well in OCM due to its high thermal stability, suitable alkaline sites, and selective mobile oxygen sites. Doping La₂Ce₂O₇ with Ca increases the alkalinity, which can considerably increase C₂ selectivity in the OCM. We have explored a scalable method of combustion synthesis, a high-throughput technique that provides favorable porous microstructures for stable catalysts at high temperatures. The obtained La₂Ce_{2-x}Ca_xO₇ powders (0 ≤ x ≤ 0.5) SEM images evidenced a microstructure composed of porous sponge-like agglomerated particles with an irregular shape, expected from the type of synthesis. The materials showed crystalline structures of disordered fluorite and C-type, indicated by both XRD and Raman analysis. Raman spectroscopy data also evidenced the presence of surface oxygen vacancies on materials, which benefited the OCM reaction. The catalytic tests showed us that the addition of Ca increased the C₂ selectivity by 10%, at 750°C and a CH₄:O₂ molar ratio of 6:1. [2]

Acknowledgments:

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References:

[1] Thyssen, V.V. et al. Chem Rev 122, 3966 (2022).

[2] Vilela, V.B. et al. ECS Trans 103, 1917 (2021).